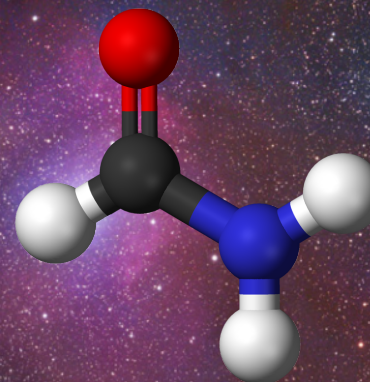
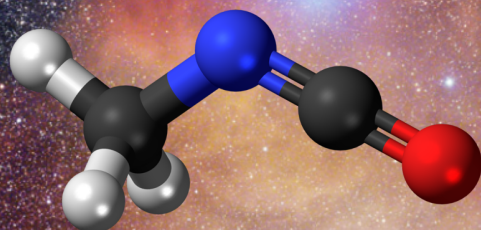


# Chemical modelling of formamide and methyl isocyanate in star-forming regions



David Quénard

Post-Doctoral Research Assistant

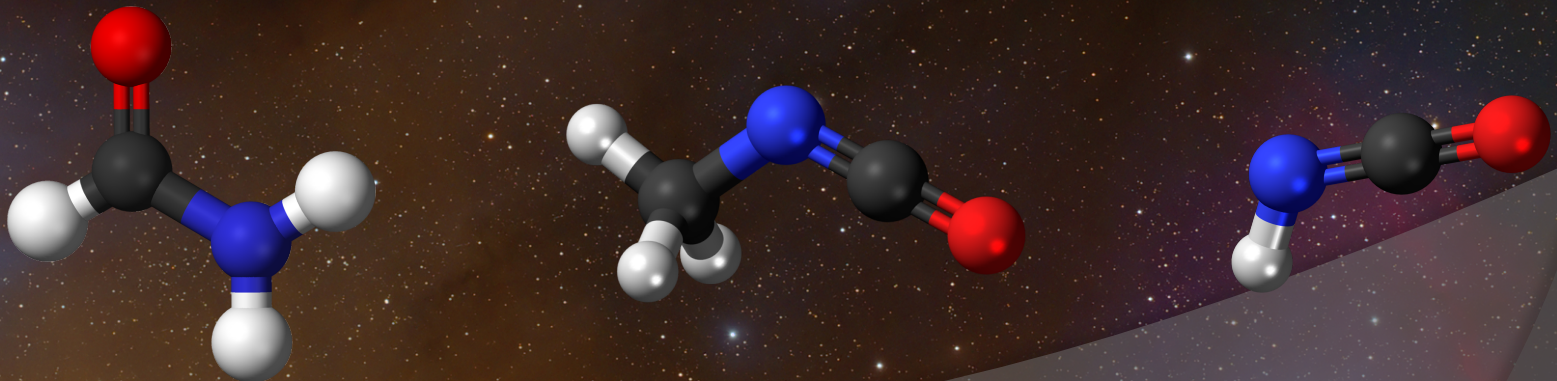
Izaskun Jiménez-Serra (QMUL), Serena Viti (UCL), Jon Holdship (UCL), Audrey Coutens (LAB)

# The search for pre-biotic species

## The peptide bond: CO-NH

Important bond in biochemistry  
(link between two amino-acid)

Several species detected with a peptide-like bond  
(e.g.  $\text{NH}_2\text{CHO}$ ,  $\text{CH}_3\text{NCO}$ ) or a peptide bond ( $\text{HNCO}$ )



# The search for pre-biotic species

Glycine  
( $\text{NH}_2\text{CH}_2\text{COOH}$ )

Methylamine  
 $\text{NH}_2\text{CH}_3$

Cyanamide  
 $\text{NH}_2\text{CN}$

Aminoacetonitrile  
 $\text{NH}_2\text{CH}_2\text{CN}$

Glycolaldehyde  
 $\text{HOCH}_2\text{CHO}$

Methyl Formate  
 $\text{HCOOCH}_3$

Acetic Acid  
 $\text{CH}_3\text{COOH}$

Hydroxylamine  
 $\text{NH}_2\text{OH}$

Acetamide  
 $\text{NH}_2\text{COCH}_3$

Formamide  
 $\text{NH}_2\text{CHO}$

N-Methyl Formamide  
 $\text{N-CH}_3\text{NHCHO}$

Isocyanic Acid (+isomers)  
 $\text{HNCO/HCNO/HOCN}$

Methyl Isocyanate  
 $\text{CH}_3\text{NCO (+isomers)}$

→ Understand the chemistry of glycine precursors and COM-related species.

# The search for pre-biotic species

**Glycine**  
**(NH<sub>2</sub>CH<sub>2</sub>COOH)**

**Methylamine**  
**NH<sub>2</sub>CH<sub>3</sub>**

**Cyanamide**  
**NH<sub>2</sub>CN**

**Aminoacetonitrile**  
**NH<sub>2</sub>CH<sub>2</sub>CN**

**Glycolaldehyde**  
**HOCH<sub>2</sub>CHO**

**Methyl Formate**  
**HCOOCH<sub>3</sub>**

**Acetic Acid**  
**CH<sub>3</sub>COOH**

**Hydroxylamine**  
**NH<sub>2</sub>OH**

**Acetamide**  
**NH<sub>2</sub>COCH<sub>3</sub>**

**Formamide**  
**NH<sub>2</sub>CHO**

**N-Methyl Formamide**  
**N-CH<sub>3</sub>NHCHO**

**Isocyanic Acid (+isomers)**  
**HNCO/HCNO/HOCN**

**Methyl Isocyanate**  
**CH<sub>3</sub>NCO (+isomers)**

**→ Understand the chemistry of glycine precursors and COM-related species.**

# Modelling of NH<sub>2</sub>CHO, CH<sub>3</sub>NCO, HNCO (& isomers)

**UCL\_CHEM** (Viti et al. 2004; Holdship et al. 2017)

<https://uclchem.github.io/>

Gas-phase + dust grain chemical code (364 species; 3446 reactions)

Recently proposed **gas-phase/grain-surface reactions**  
for **HNCO and CH<sub>3</sub>NCO** (+ isomers)

**# = grain surface**

Reactions	Reference
Isocyanic Acid – HNCO/HOCN/HCNO	
Complex gas/grain network	Quan et al. (2010)
$\#NH + \#CO \longrightarrow \#HNCO$	Fedoseev et al. (2015)
Methyl Isocyanate – CH <sub>3</sub> NCO	
$HNCO + CH_3 \longrightarrow CH_3NCO + H$	Halfen et al. (2015)
$\#CH_3 + \#OCN \longrightarrow \#CH_3NCO$	Belloche et al. (2017); Ligterink et al. (2017)
$\#CH_3 + \#HNCO \longrightarrow \#CH_3NCO + \#H$	Ligterink et al. (2017)
$\#CH_3 + \#HNCO \longrightarrow \#CH_4 + \#OCN$	Ligterink et al. (2017)
$\#CH_3NCO + \#H \longrightarrow \#CH_3NH + \#CO$	Ligterink et al., private communication

# Modelling of NH<sub>2</sub>CHO, CH<sub>3</sub>NCO, HNCO (& isomers)

## Recently proposed **gas-phase/grain-surface reactions** for NH<sub>2</sub>CHO

Reactions	Reference
Formamide – NH <sub>2</sub> CHO	
NH <sub>2</sub> + H <sub>2</sub> CO → NH <sub>2</sub> CHO + H	Skouteris et al. (2017)
#HNCO + #H → #NH <sub>2</sub> + #CO	Song & Kästner (2016)
#HNCO + #H → #H <sub>2</sub> NCO	Song & Kästner (2016)
#H <sub>2</sub> NCO + #H → #NH <sub>2</sub> CHO	Song & Kästner (2016)
#H <sub>2</sub> NCO + #H → #HNCO + #H <sub>2</sub>	Noble et al. (2016)
#NH <sub>2</sub> + #HCO → #NH <sub>2</sub> CHO	Fedoseev et al. (2016)
#NH <sub>2</sub> + #HCO → #NH <sub>3</sub> + CO	Fedoseev et al. (2016)
#NH <sub>2</sub> + #H <sub>2</sub> CO → #NH <sub>2</sub> CHO + #H	Fedoseev et al. (2016)
#NH <sub>2</sub> + #H <sub>2</sub> CO → #NH <sub>3</sub> + #HCO	Fedoseev et al. (2016)
#H <sub>2</sub> NCO + #CH <sub>3</sub> → #CH <sub>3</sub> CONH <sub>2</sub>	Belloche et al. (2017)
#NH <sub>2</sub> CHO + #OH → #H <sub>2</sub> NCO + #H <sub>2</sub> O	Belloche et al. (2017)
#NH <sub>2</sub> CHO + #CH <sub>2</sub> → #CH <sub>3</sub> CONH <sub>2</sub>	Belloche et al. (2017)

# COMs in the pre-stellar core L1544

O-bearing and N-bearing COMs are **more abundant at  $r \sim 4000$  AU** (methanol peak position) (Jiménez-Serra et al. 2016)

## Important non-detections:

### Core centre

$$X[\text{NH}_2\text{CHO}] < 2.4 \times 10^{-13}$$

$$X[\text{CH}_3\text{NCO}] < 2.0 \times 10^{-12}$$

### Methanol peak

$$X[\text{NH}_2\text{CHO}] < 6.7 \times 10^{-13}$$

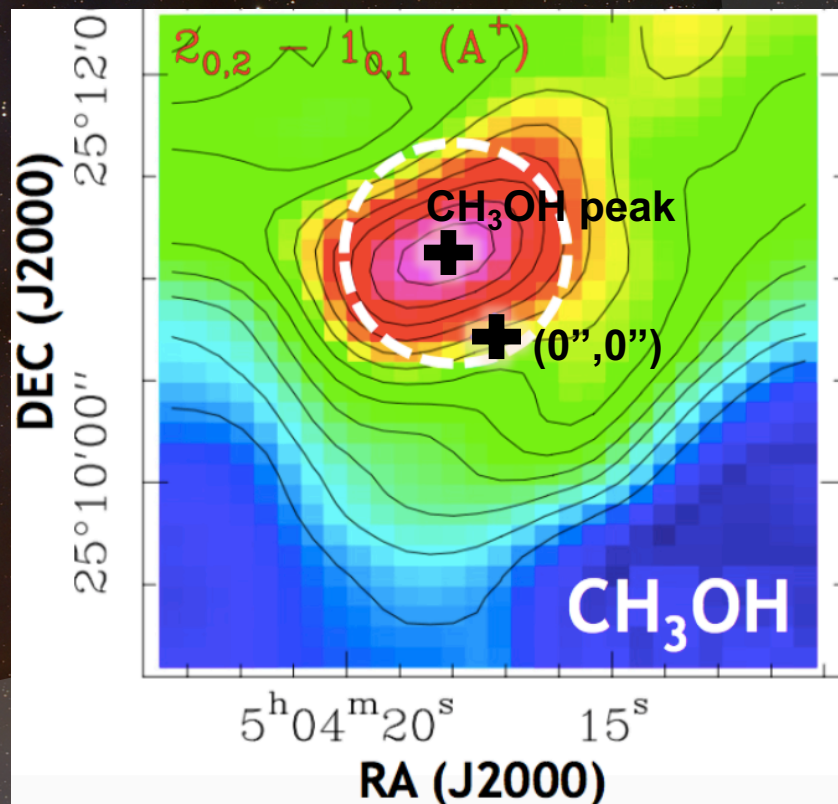
$$X[\text{CH}_3\text{NCO}] < 6.0 \times 10^{-12}$$

## Chemical modelling by Vasyunin et al. (2017)

- Gas-phase + dust grain model of Vasyunin & Herbst (2013)
- Focused on the O-bearing COMs chemical modelling

**Large discrepancy found for  $\text{NH}_2\text{CHO}$ :  $\sim 100$  times higher**

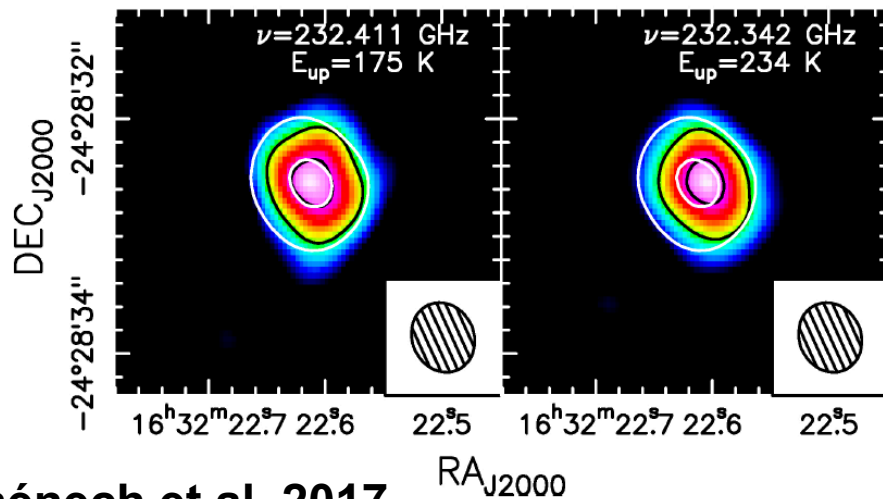
Bizzocchi et al. (2014)



# CH<sub>3</sub>NCO in IRAS16293

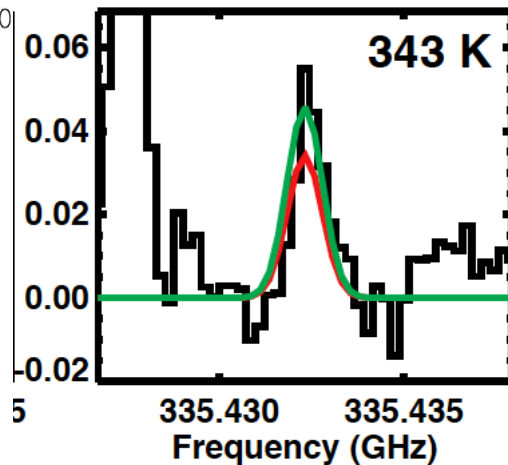
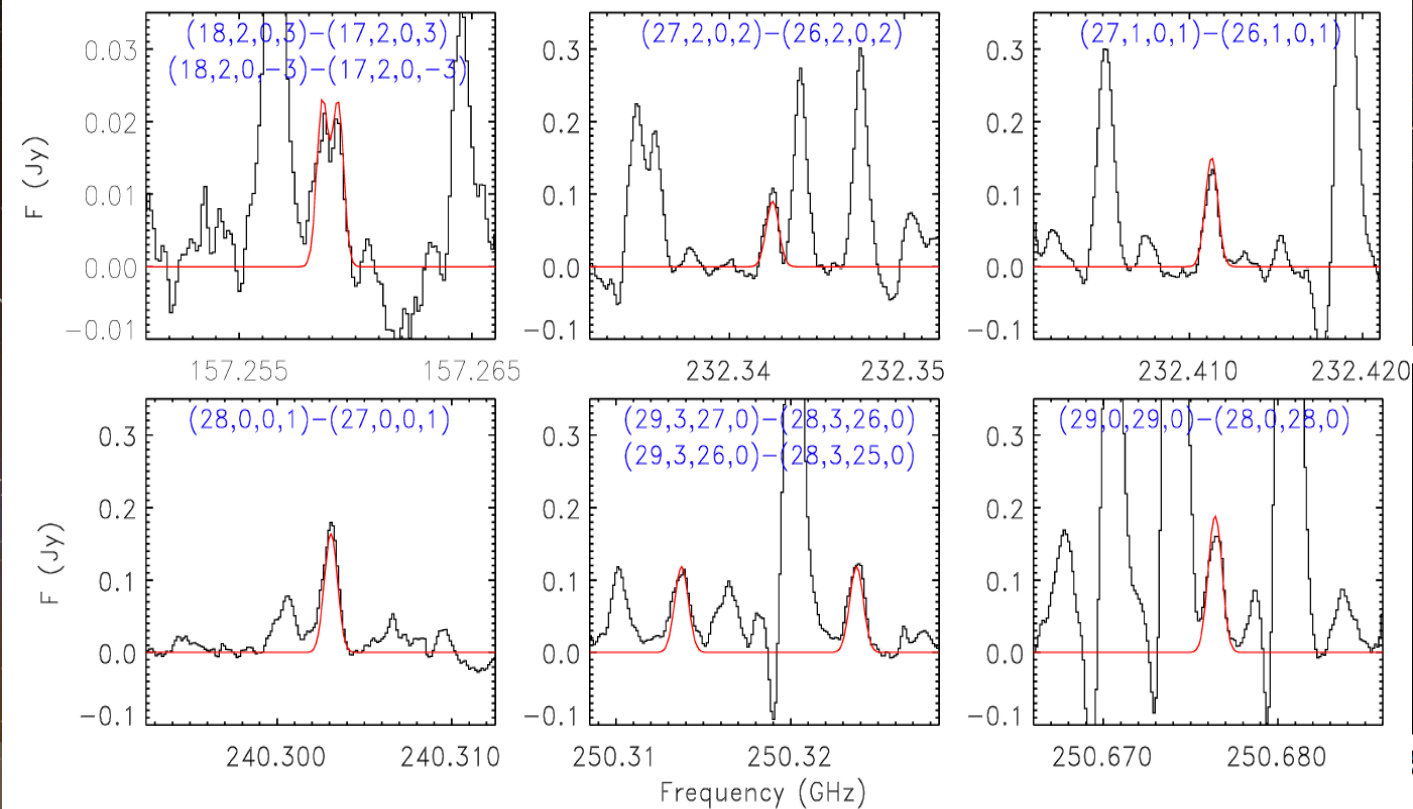
Detection of **CH<sub>3</sub>NCO**  
towards source B!

(Martín-Doménech et al. 2017,  
Ligterink et al. 2017)



Martín-Doménech et al. 2017

Ligterink et al. 2017





# N-bearing COMs chemical modelling in L1544

## 2-steps chemical modelling

### *Phase 0*

- Diffuse cloud step with  $n_{\text{H}} = 100 \text{ cm}^{-3}$  and  $T=20 \text{ K}$ .
- Evolution of the chemistry for a few millions years.
- Low  $A_{\text{V}}$ : no icy mantle formation but gas phase chemistry.

### *Phase 1*

- Collapse phase to  $n_{\text{H}} = 5 \times 10^6 \text{ cm}^{-3}$  (core centre) and  $n_{\text{H}} = 4 \times 10^5 \text{ cm}^{-3}$  (methanol peak) with  $T=10 \text{ K}$ .
- Depletion of species onto grain surface

# N-bearing COMs chemical modelling in IRAS16293

## 3-steps chemical modelling

### *Phase 0*

→ Same as for L1544 (diffuse cloud phase)

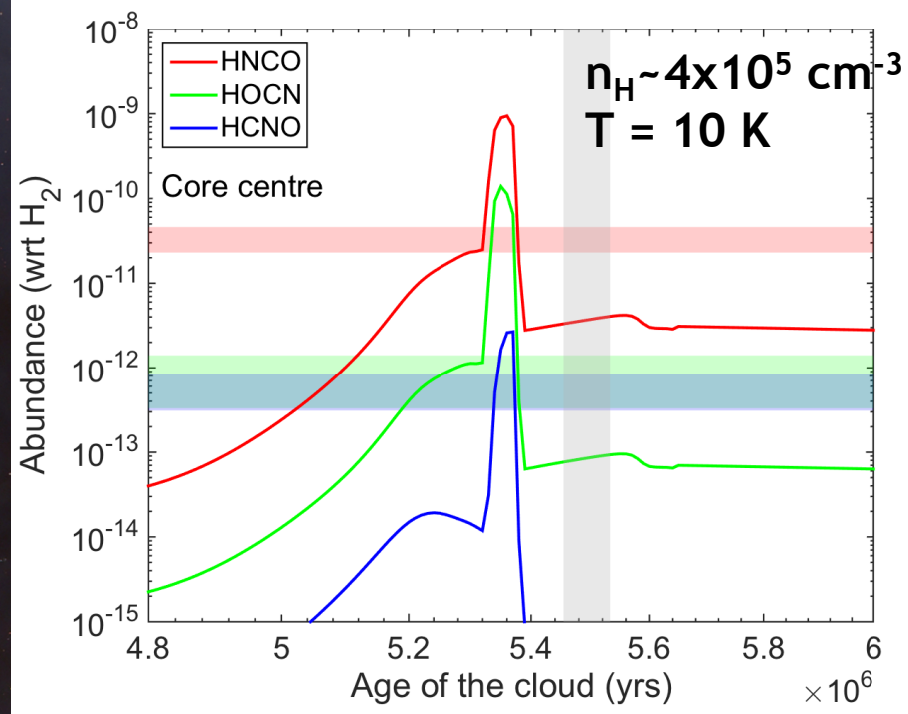
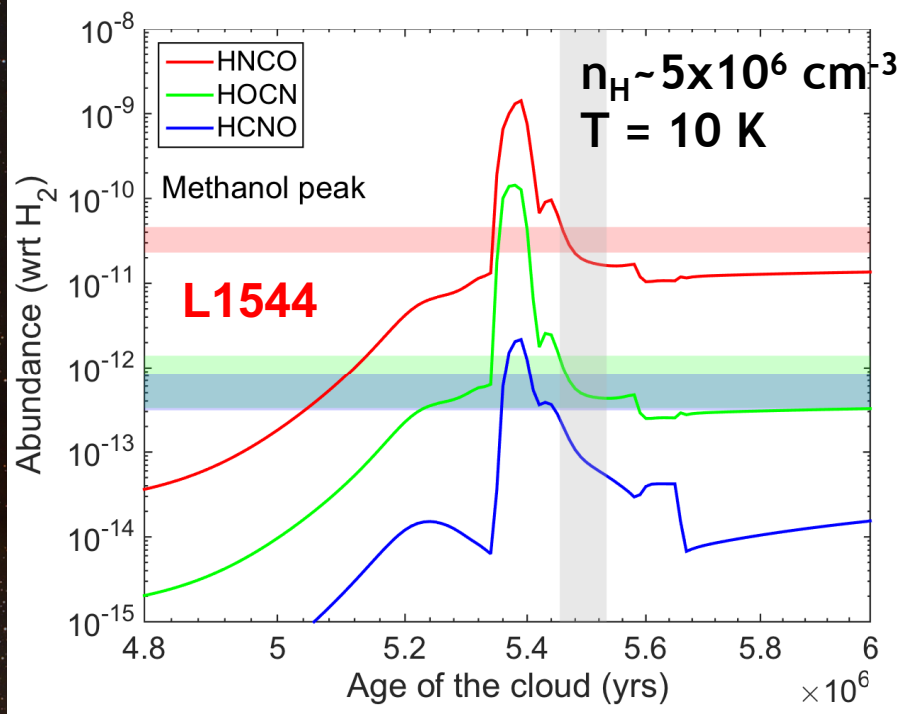
### *Phase 1*

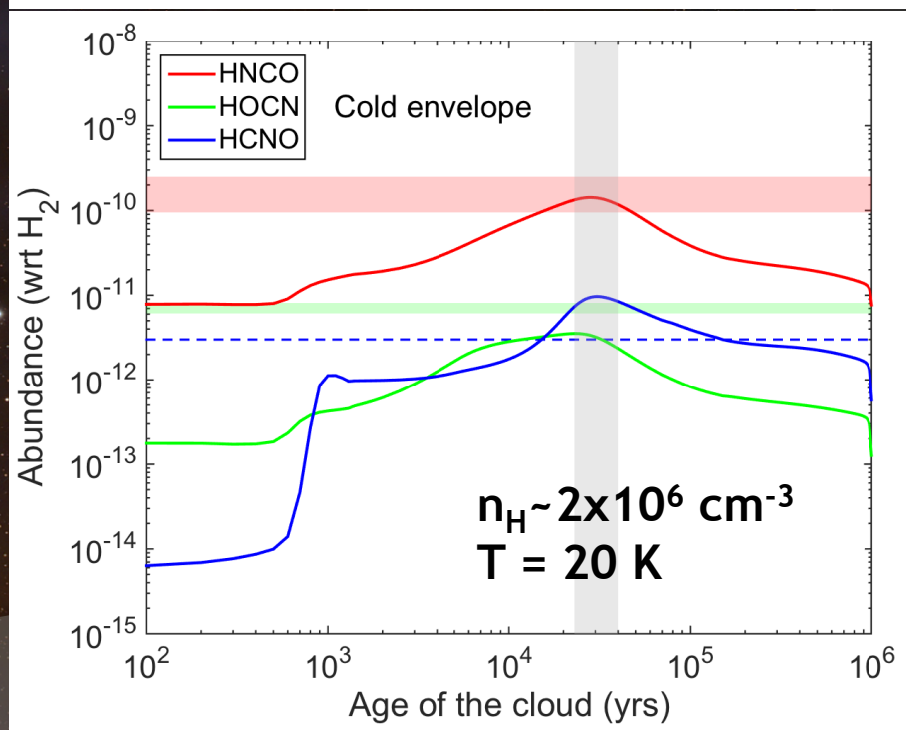
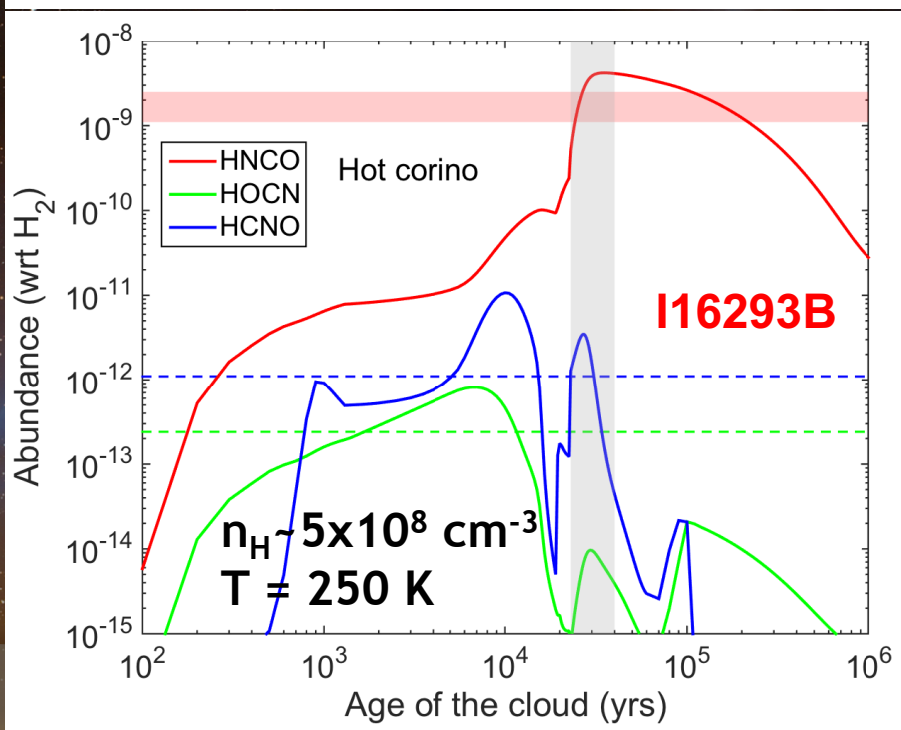
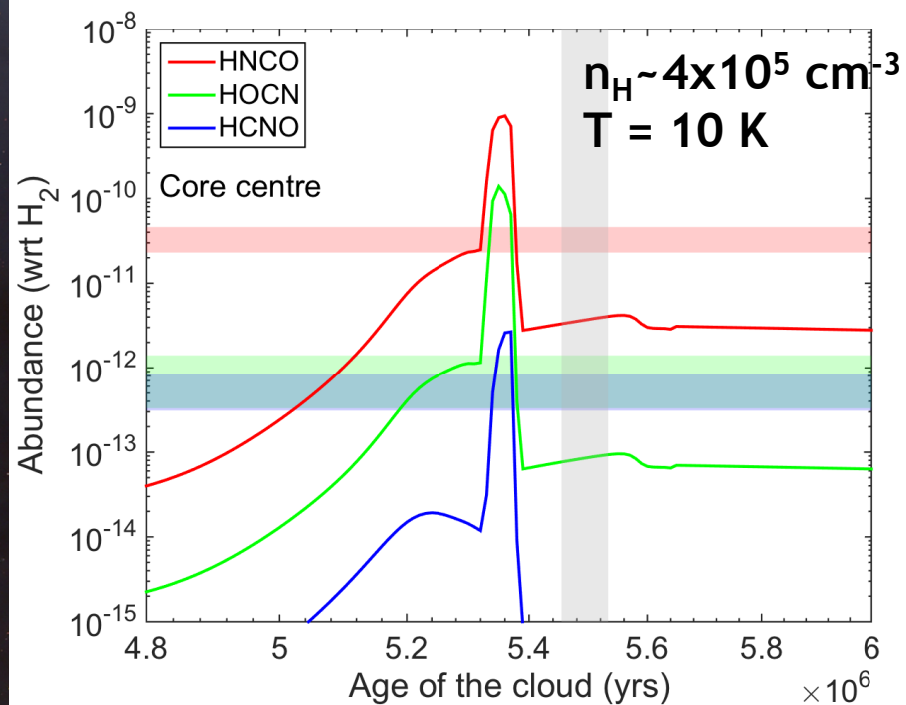
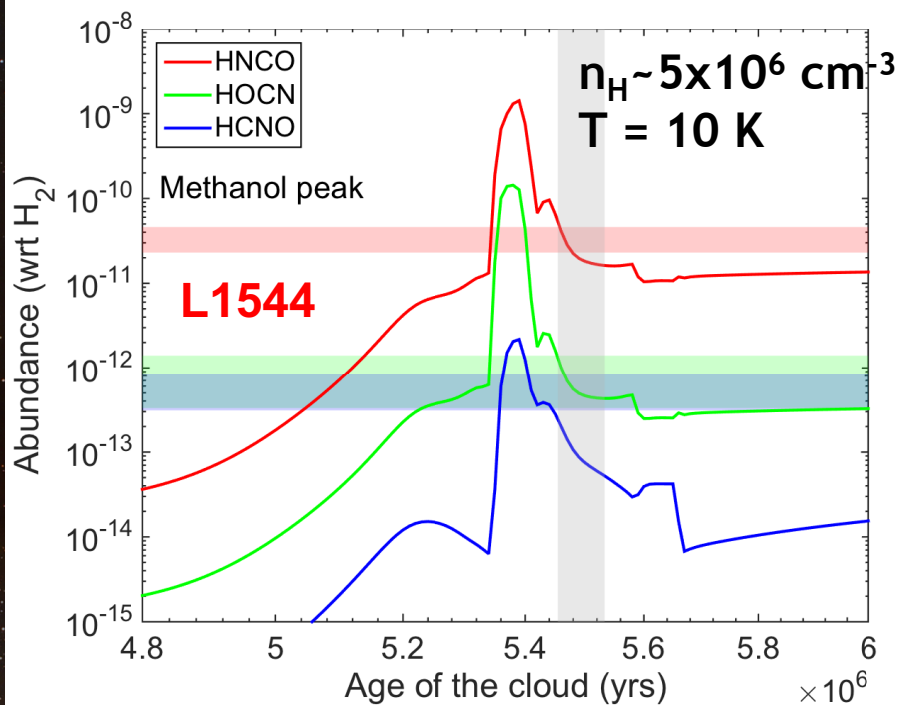
→ Collapse phase to  $n_{\text{H}} = 5 \times 10^8 \text{ cm}^{-3}$  (hot corino) and  $n_{\text{H}} = 2 \times 10^6 \text{ cm}^{-3}$  (cold envelope) with  $T=10 \text{ K}$ .

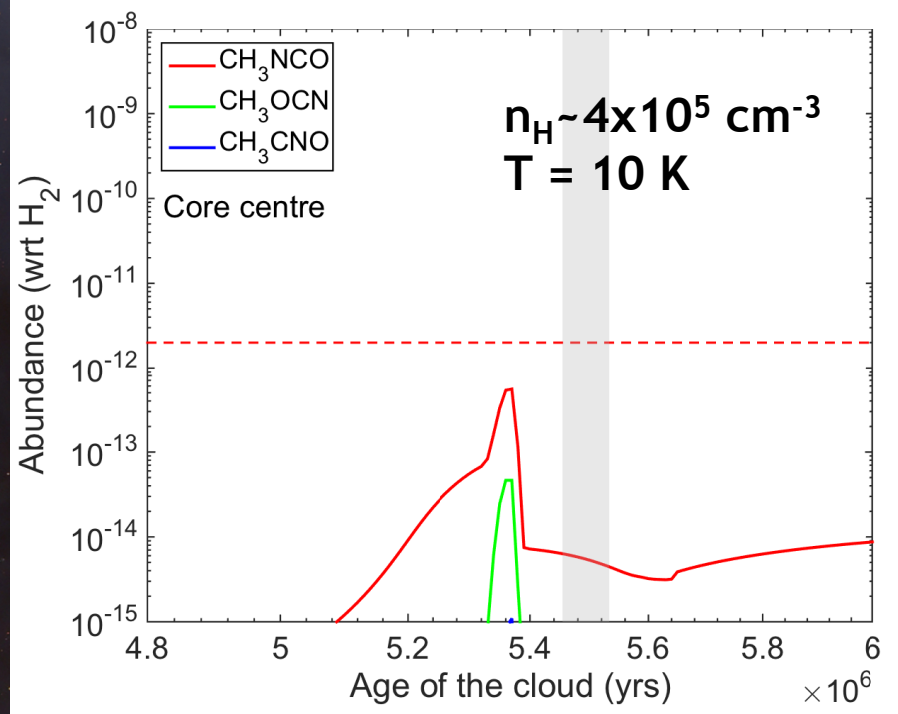
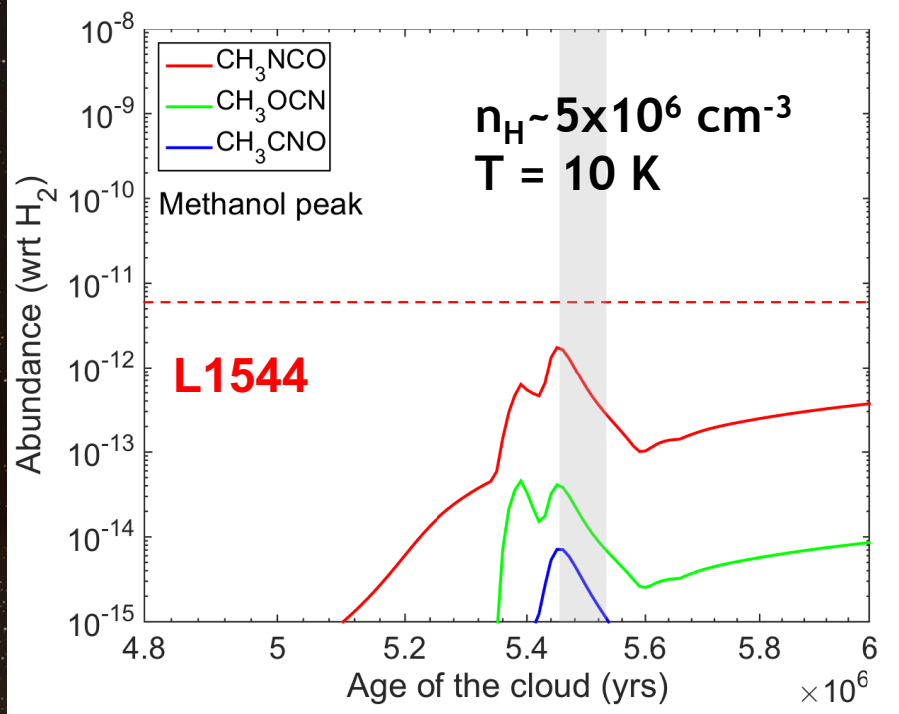
### *Phase 2*

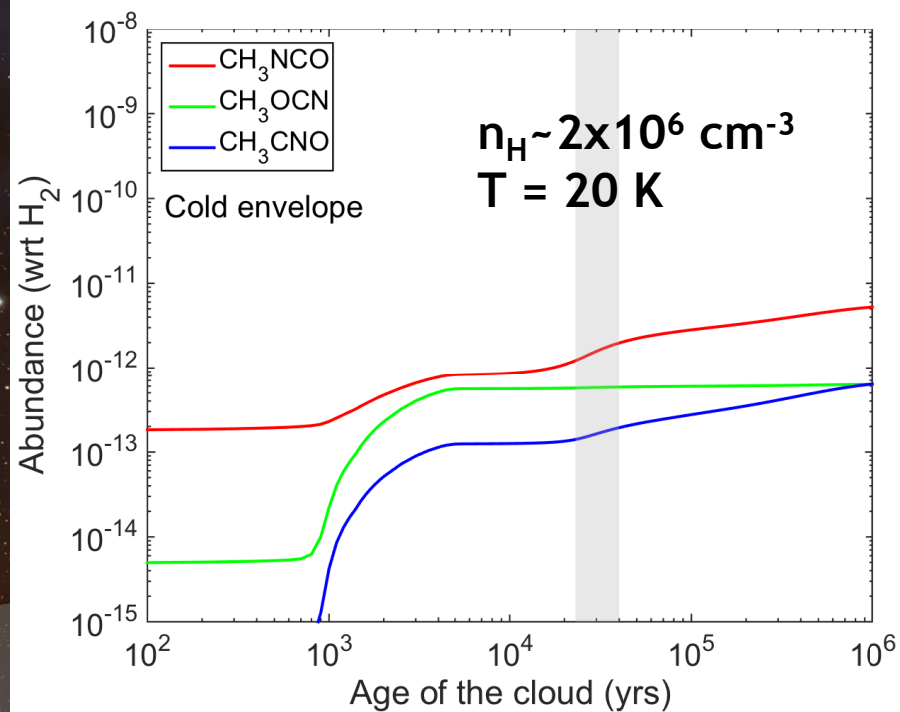
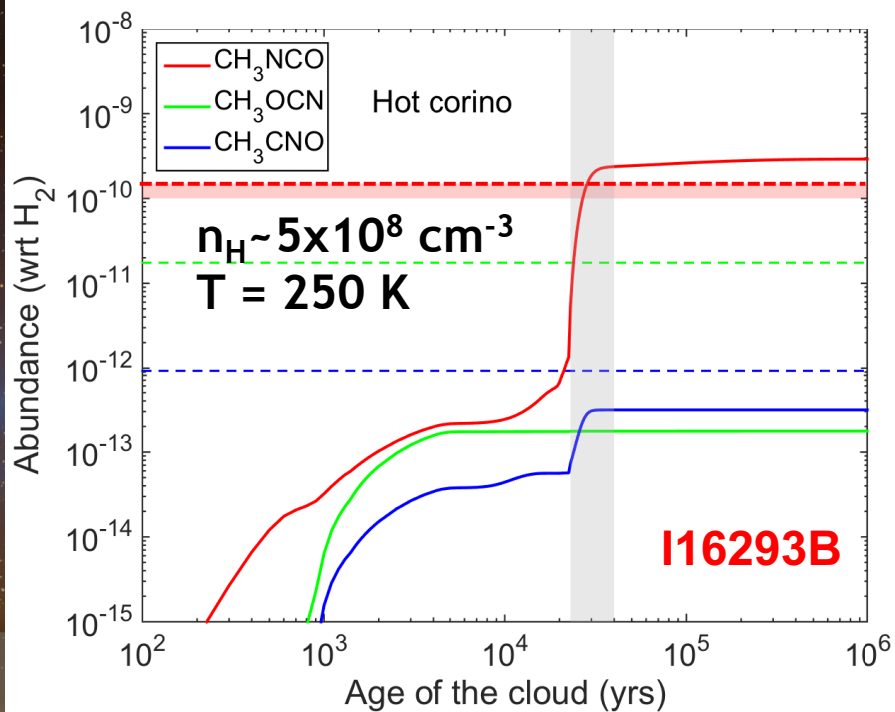
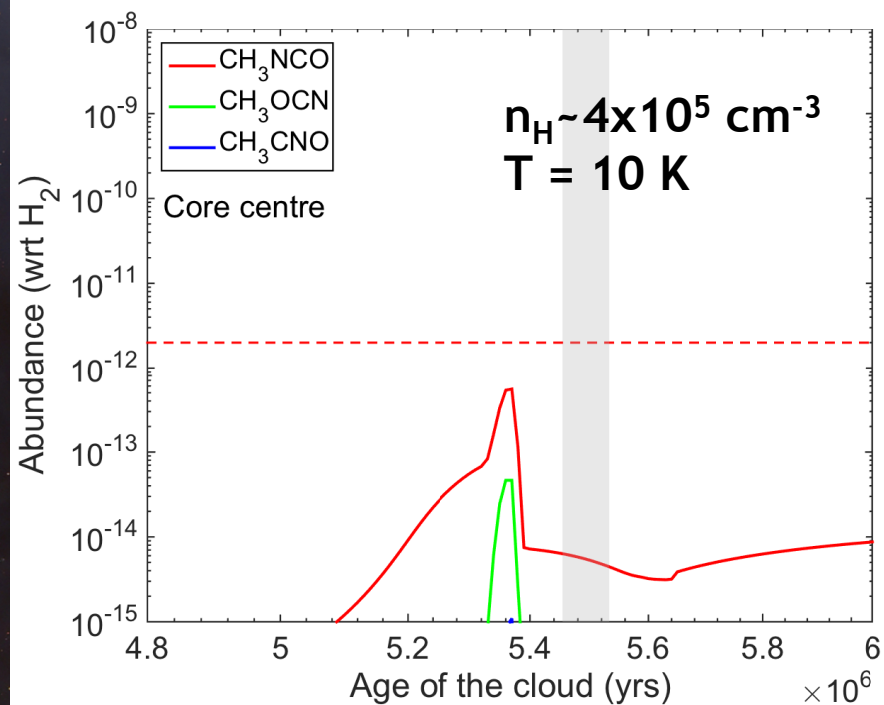
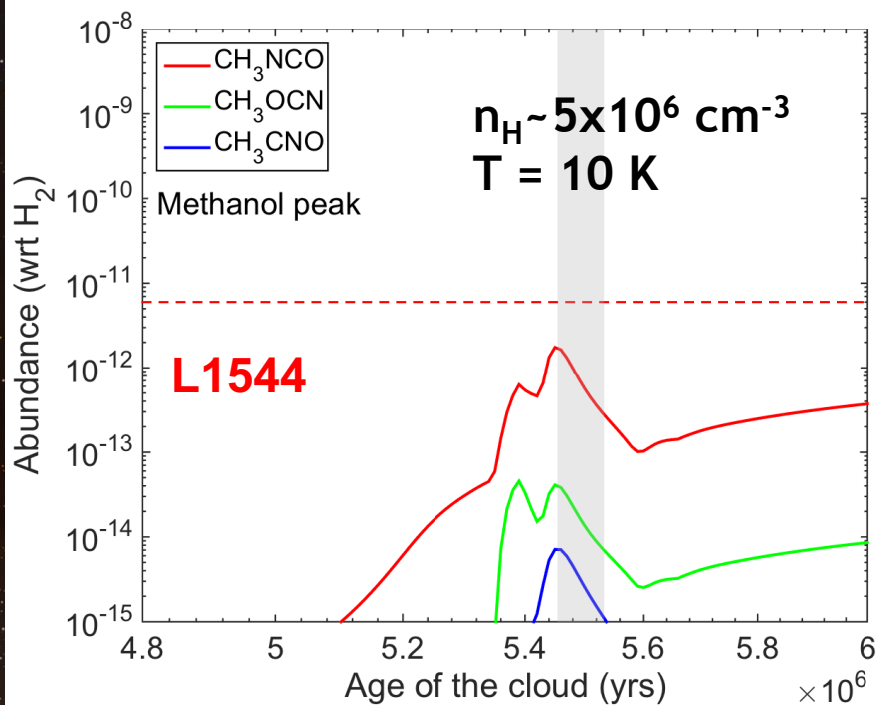
→ Warm-up phase to 250 K (hot corino) and 20 K (cold envelope)

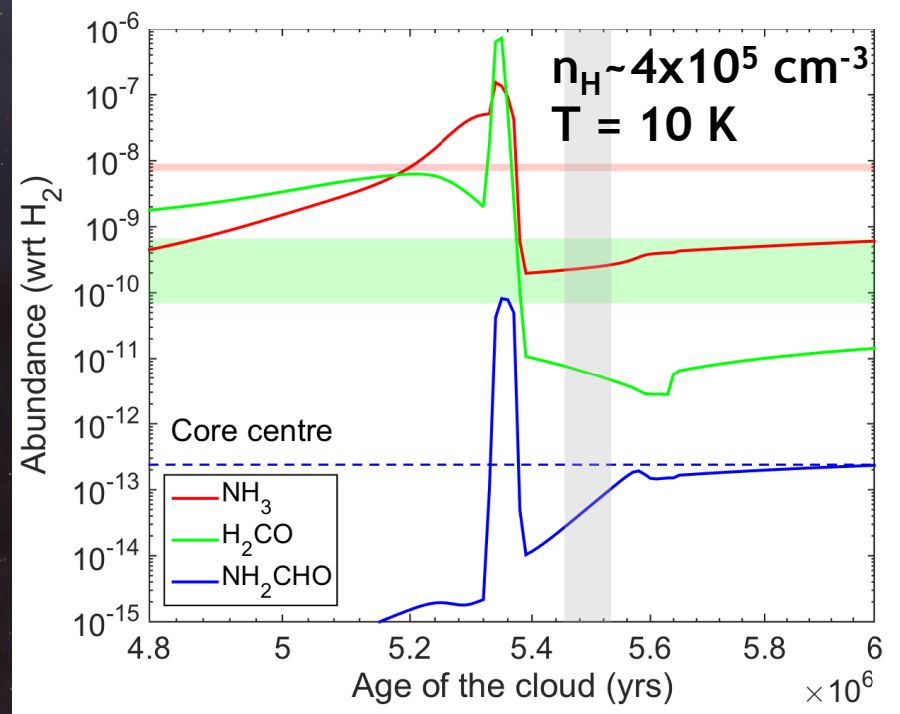
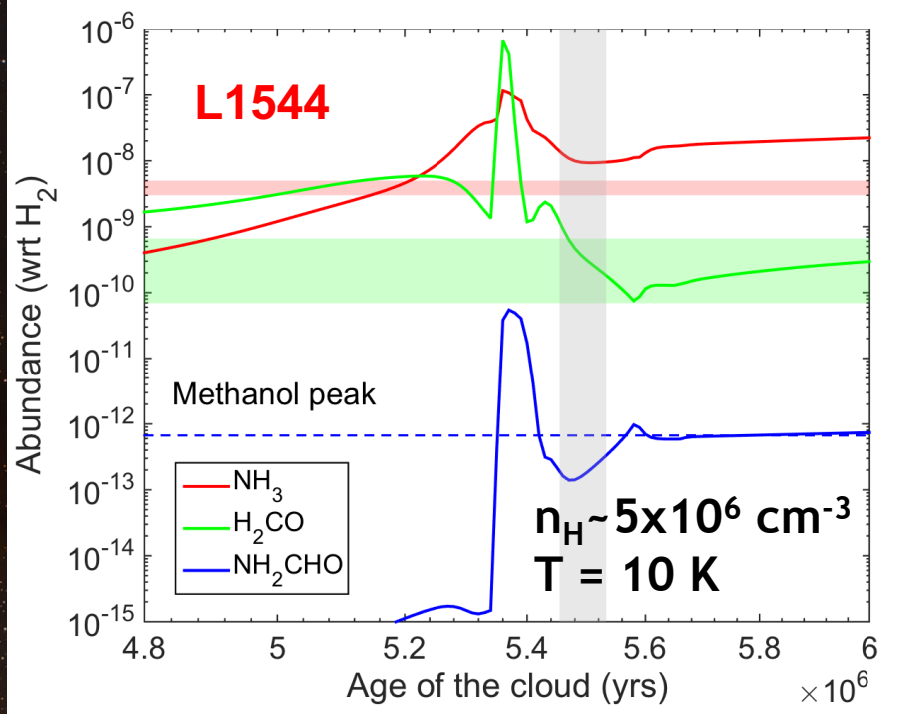
→ Desorption of species from the grain mantle at high T !

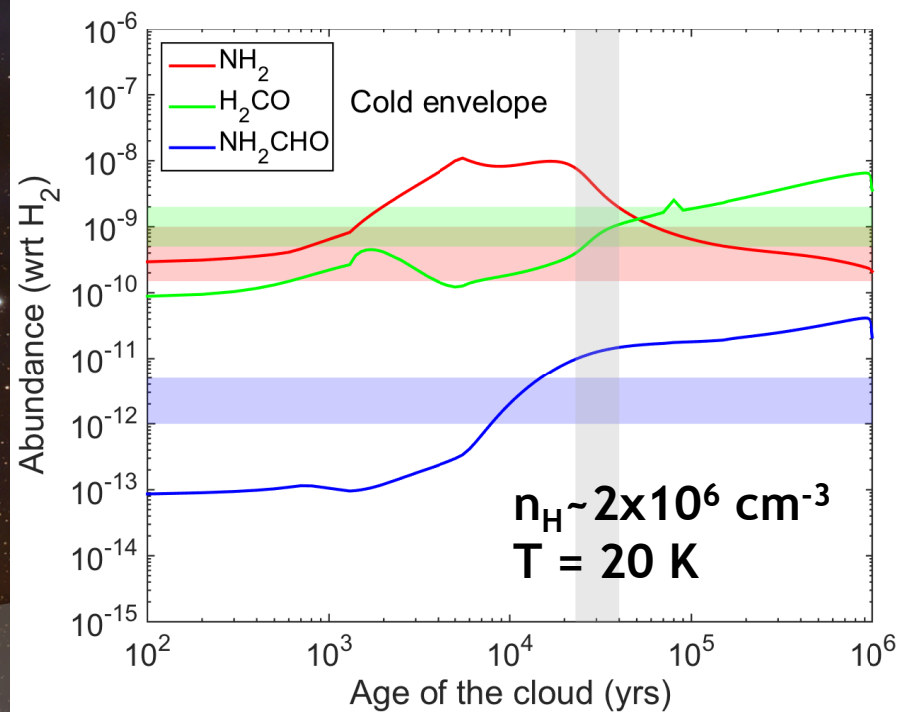
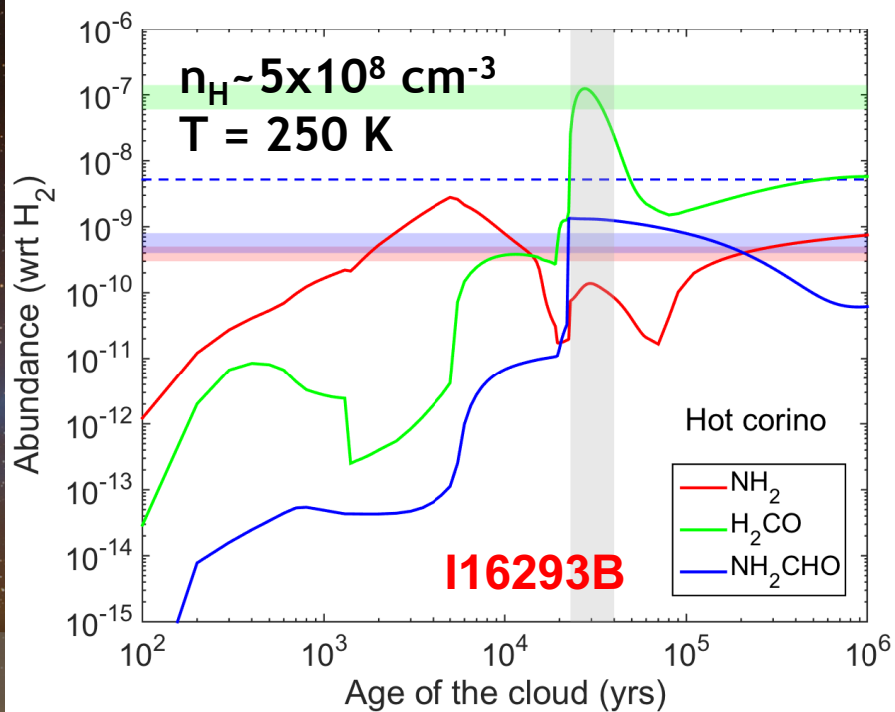
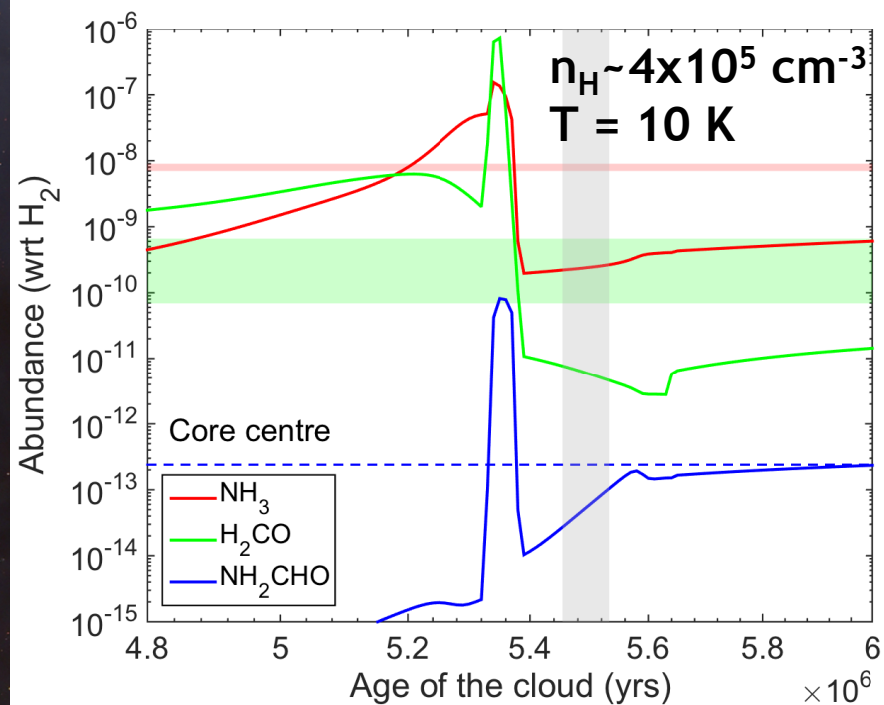
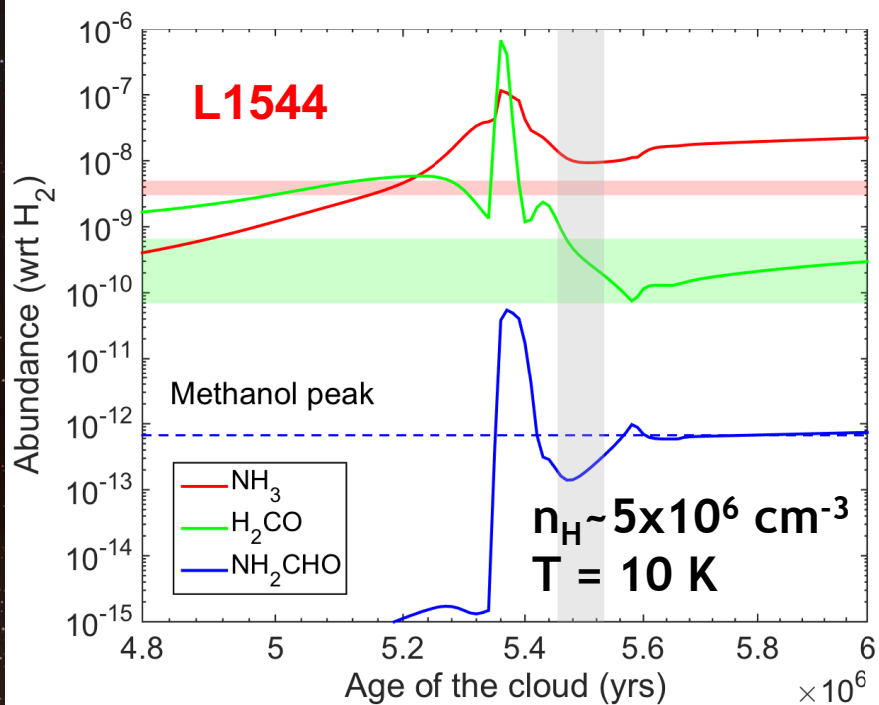










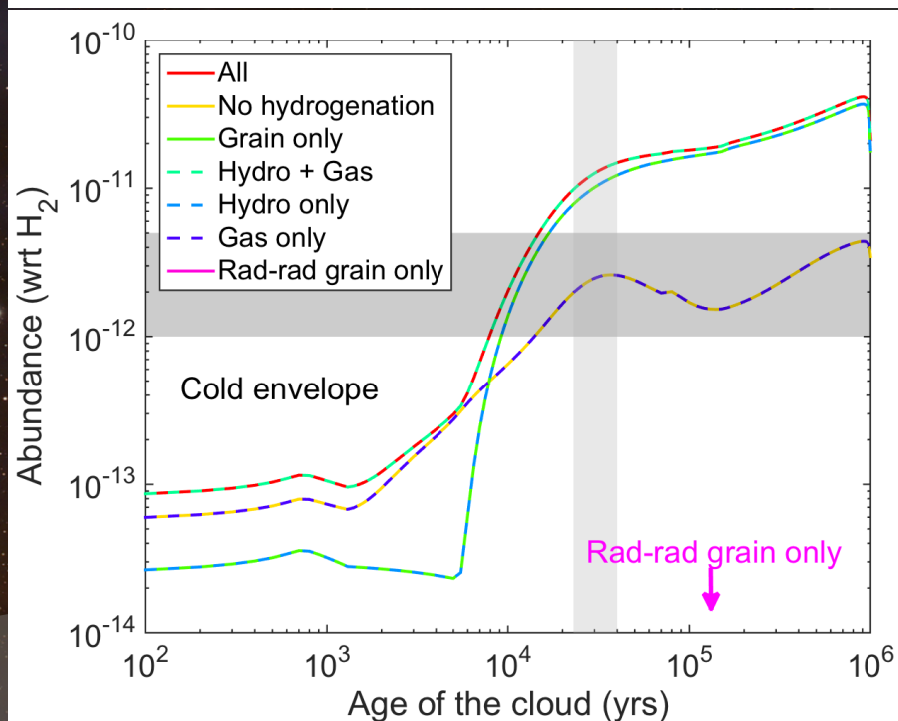
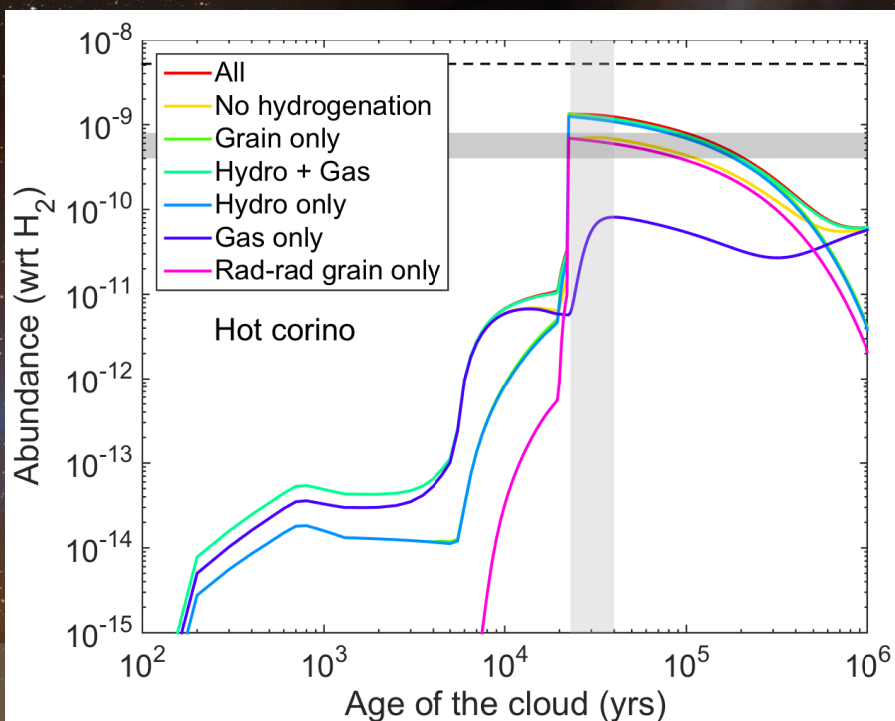
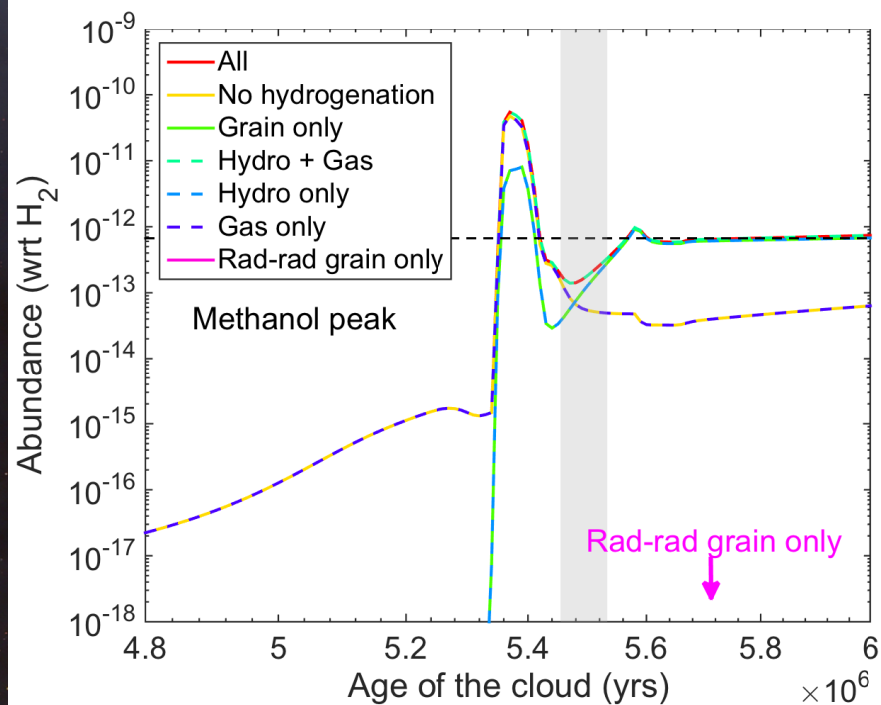






# NH<sub>2</sub>CHO chemistry

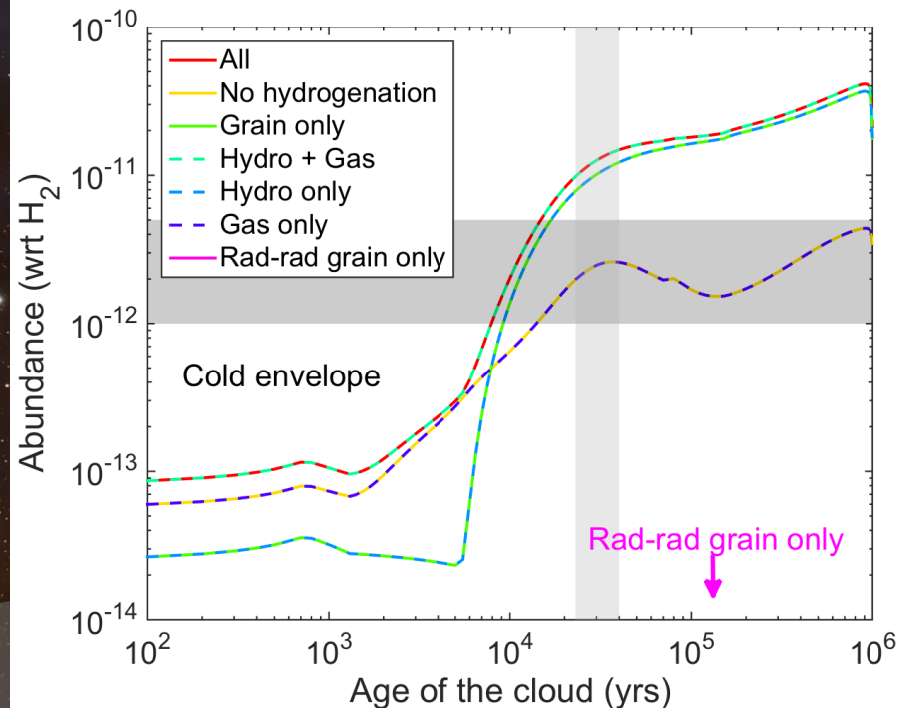
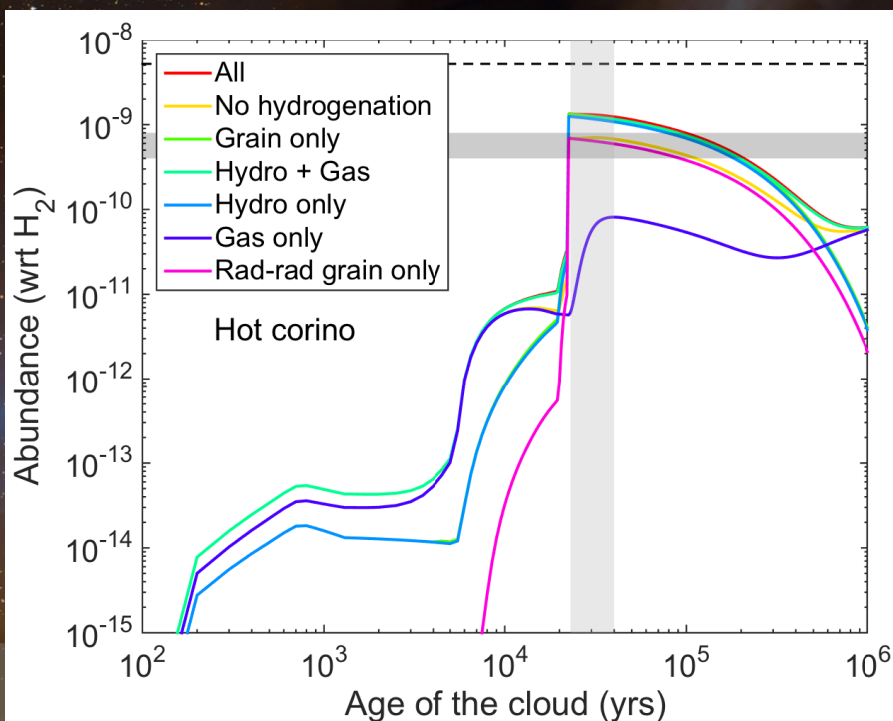
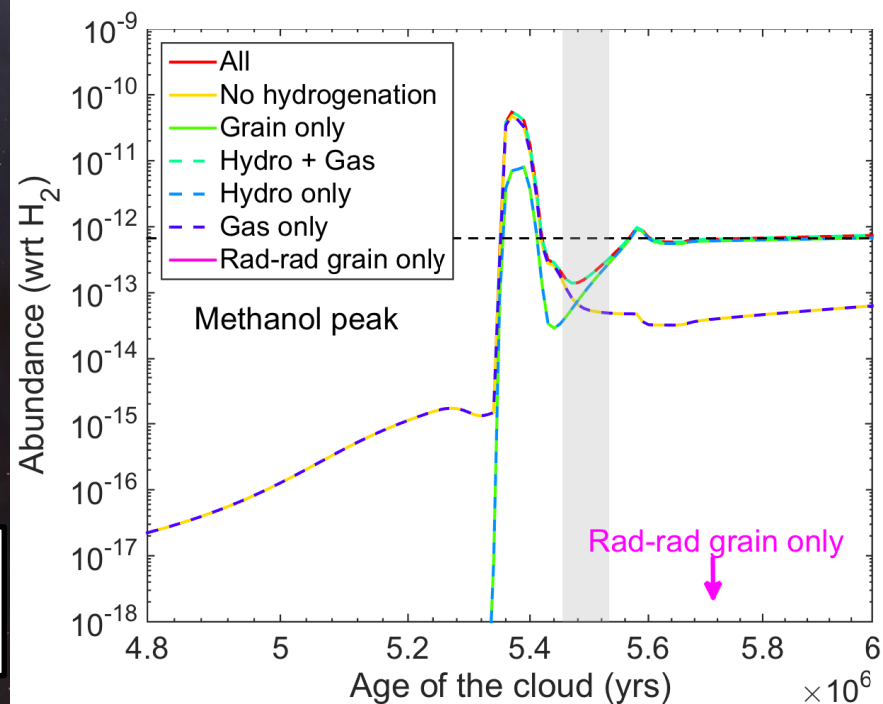
- Gas phase chemistry
- Grain surface chemistry:
  - Radical-radical reactions
  - Hydrogenation



# NH<sub>2</sub>CHO chemistry

- Gas phase chemistry
- Grain surface chemistry:
  - Radical-radical reactions
  - Hydrogenation

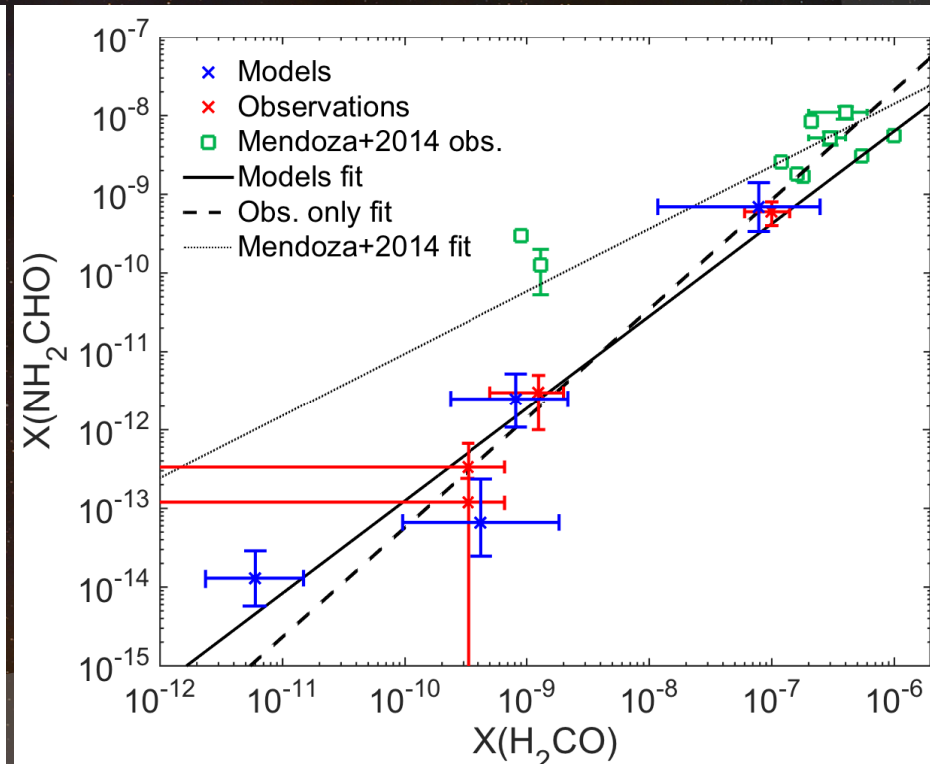
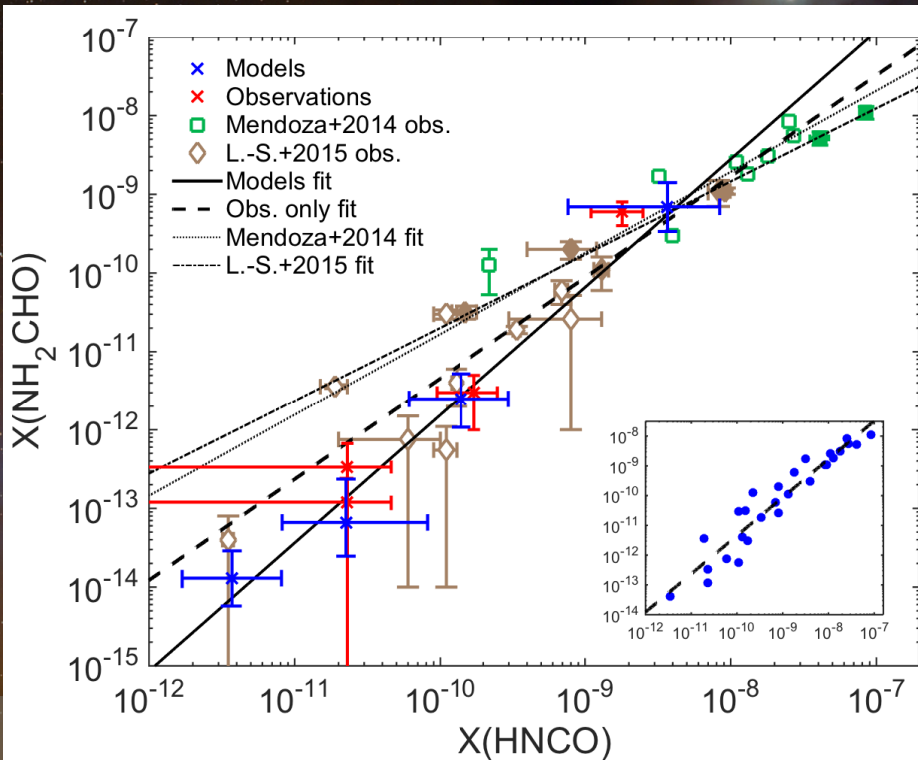
**Modelling different physical regimes help to constrain the chemistry !**



# HNCO & H<sub>2</sub>CO vs NH<sub>2</sub>CHO

**Mendoza et al. (2014)** and **López-Sepulcre et al. (2015)**:  
Observational correlation → **Chemical correlation** between the two?

Modelling of NH<sub>2</sub>CHO (no hydrogenation from HNCO)  
→ **Physical (environmental) correlation** depending mainly on the  
**temperature** that triggers different **chemical processes**.



# Conclusions

- Modelling of N-bearing COMs predicts abundances of **NH<sub>2</sub>CHO**, **CH<sub>3</sub>NCO** (and isomers), **HNCO** (and isomers) in L1544 and IRAS16293 B
  - L1544: **core centre and methanol peak**
  - IRAS16293 B: **hot corino and cold envelope**
- Both **gas-phase and grain-phase chemistry** are needed to explain the observed abundances of NH<sub>2</sub>CHO
  - Hydrogenation of HNCO tend to overestimate the NH<sub>2</sub>CHO abundance compared to radical-radical reactions
- The observed correlation between **HNCO and NH<sub>2</sub>CHO** may come from an **environmental correlation (temperature)** rather than a chemical correlation